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(71) Applicant: NIPPONDENSO CO., LTD.  
Kariya-city Aichi-pref., 448 (JP)

(72) Inventors:  
• Kondo, Toshiharu,  
c/o Nippondenso Co., Ltd.  
Kariya-city, Aichi-pref. 448 (JP)

• Ito, Keiji,  
c/o Nippondenso Co., Ltd.  
Kariya-city, Aichi-pref. 448 (JP)  
• Kageyama, Terutaka,  
c/o Nippondenso Co., Ltd.  
Kariya-city, Aichi-pref. 448 (JP)  
• Obata, Takashi,  
c/o Nippondenso Co., Ltd.  
Kariya-city, Aichi-pref. 448 (JP)

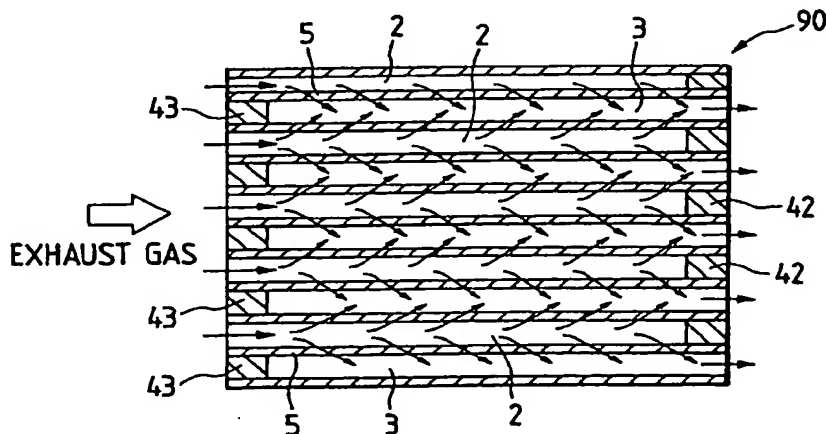
(74) Representative: KUHNEN, WACKER & PARTNER  
Alois-Steinecker-Strasse 22  
85354 Freising (DE)

### (54) Honeycomb structural body and method of manufacturing the same

(57) A honeycomb structural body has cordierite as a main component. The cordierite has a chemical composition including 45-55% SiO<sub>2</sub>, 33-42% Al<sub>2</sub>O<sub>3</sub>, and 12-18% MgO by weight. The honeycomb structural body has a thermal expansion coefficient equal to or less than  $0.3 \times 10^{-6}/^{\circ}\text{C}$  from 25° C to 800° C. The honeycomb structural body has a porosity of 55-80%. The honeycomb structural body has a mean pore diameter

of 25-40 μm. The honeycomb structural body includes partition walls. Pores at surfaces of the partition walls include small pores having diameters of 5-40 μm and large pores having diameters of 40-100 μm. The number of the small pores equals five to forty times that of the large pores.

FIG. 2



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## Description

This invention relates to a cordierite honeycomb structural body which is usable in a filter for catching particulates emitted from a diesel engine. This invention also relates to a method of manufacturing such a cordierite honeycomb structural body.

Some filters for catching particulates emitted from diesel engines use cordierite honeycomb structural bodies.

As shown in Figs. 1 and 2, a prior-art honeycomb structure body 90 has a cylindrical shape. The prior-art honeycomb structural body 90 is formed with a lot of axially-extending inlet passages 2 and outlet passages 3.

As shown in Figs. 1 and 2, upstream ends of the inlet passages 2 are open to introduce engine exhaust gas thereinto. On the other hand, downstream ends of the inlet passages 2 are blocked or closed by choking members 42. Upstream ends of the outlet passages 3 are blocked or closed by choking members 43 while downstream ends thereof are open. As shown in Figs. 1 and 2, the inlet passages 2 and the outlet passages 3 alternate with each other in a chequered pattern along a vertical direction and a horizontal direction.

Partition walls 5 defining the inlet passages 2 and the outlet passages 3 are porous. Thus, the partition walls 5 have a lot of pores.

A filter using the prior-art honeycomb structural body 90 catches particulates as follows. With reference to Fig. 2, engine exhaust gas containing particulates enters the inlet passages 2. Then, the engine exhaust gas moves from the inlet passages 2 to the outlet passages through pores in the partition walls 5 since the downstream ends of the inlet passages 2 are closed. At this time, the partition walls 5 catch particulates, and thereby cleans the engine exhaust gas. Subsequently, the cleaned engine exhaust gas flows along the outlet passages 3 before exiting from the downstream ends thereof.

It is desirable that such a filter using a honeycomb structural body meets three important requirements, that is, a requirement for a high efficiency of catch of particulates (a high filtration efficiency), a requirement for a low pressure loss, and a requirement for a low coefficient of thermal expansion.

Japanese published unexamined patent application 5-254958 corresponding to United States Patent 5,258,150 discloses a cordierite honeycomb body having a coefficient of thermal expansion of no greater than about  $4 \times 10^{-7}/^{\circ}\text{C}$  from about  $25^{\circ}\text{C}$  to about  $800^{\circ}\text{C}$ , and a total porosity of greater than about 42%.

The cordierite honeycomb body in Japanese application 5-254958 has a low coefficient of thermal expansion. Accordingly, the cordierite honeycomb body in Japanese application 5-254958 is hardly damaged by thermal expansion. The cordierite honeycomb body in Japanese application 5-254958 has a high filtration efficiency. However, it appears that the cordierite honey-

comb body in Japanese application 5-254958 has a low porosity. The low porosity causes a great pressure loss.

None of other prior-art honeycomb structural bodies meets all the previously-indicated three requirements. Specifically, the other prior-art honeycomb structural bodies meet two of the three requirements but do not meet the remaining requirement.

This invention has been carried out in view of such a problem in the prior-art honeycomb structural bodies. It is a first object of this invention to provide a honeycomb structural body which has a high filtration efficiency, a low pressure loss, and a low thermal expansion coefficient. It is a second object of this invention to provide a method of manufacturing such a honeycomb structural body.

A first aspect of this invention provides a honeycomb structural body having cordierite as a main component, the cordierite having a chemical composition including 45-55%  $\text{SiO}_2$ , 33-42%  $\text{Al}_2\text{O}_3$ , and 12-18%  $\text{MgO}$  by weight, the honeycomb structural body having a thermal expansion coefficient equal to or less than  $0.3 \times 10^{-6}/^{\circ}\text{C}$  from  $25^{\circ}\text{C}$  to  $800^{\circ}\text{C}$ , the honeycomb structural body having a porosity of 55-80%, the honeycomb structural body having a mean pore diameter of 25-40  $\mu\text{m}$ , the honeycomb structural body including partition walls, wherein pores at surfaces of the partition walls include small pores having diameters of 5-40  $\mu\text{m}$  and large pores having diameters of 40-100  $\mu\text{m}$ , and wherein the number of the small pores equals five to forty times that of the large pores.

It is more preferable that the porosity is in the range of 62-75%.

A second aspect of this invention provides a method of manufacturing a honeycomb structural body having cordierite as a main component, the cordierite having a chemical composition including 45-55%  $\text{SiO}_2$ , 33-42%  $\text{Al}_2\text{O}_3$ , and 12-18%  $\text{MgO}$  by weight, the honeycomb structural body having a thermal expansion coefficient equal to or less than  $0.3 \times 10^{-6}/^{\circ}\text{C}$  from  $25^{\circ}\text{C}$  to  $800^{\circ}\text{C}$ , the honeycomb structural body having a porosity of 55-80%, the honeycomb structural body having a mean pore diameter of 25-40  $\mu\text{m}$ , the honeycomb structural body including partition walls, wherein pores at surfaces of the partition walls include small pores having diameters of 5-40  $\mu\text{m}$  and large pores having diameters of 40-100  $\mu\text{m}$ , and wherein the number of the small pores equals five to forty times that of the large pores, the method comprising the steps of preparing talc containing 0.1-0.6%  $\text{Fe}_2\text{O}_3$  and 0.35% or less  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight; preparing aluminum hydroxide containing small grains having diameters of 0.5-3  $\mu\text{m}$  and large grains having diameters of 5-15  $\mu\text{m}$ , the small grains and the large grains occupying 50-100% of all grains, wherein the ratio in weight% of the large grains to the small grains is in the range of 5/95 to 95/5; preparing fused silica containing 0.01% or less  $(\text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight, and having a mean grain diameter of 30-100  $\mu\text{m}$ ; using the talc, the aluminum hydroxide, and the fused silica as at least part of raw materials, and

mixing the talc, the aluminum hydroxide, and the fused silica into a first mixture; adding at least one of organic blowing agent and combustible substance to the first mixture, the organic blowing agent starting to foam at 100° C or lower, the combustible substance starting to burn at lower than a sintering temperature, wherein the amount of added one of the organic blowing agent and the combustible substance is equal to 5-50 weight% with respect to the raw materials; kneading and mixing the first mixture and at least one of the organic blowing agent and combustible substance into a second mixture; making the second mixture into a honeycomb shape; and sintering the second mixture in the honeycomb shape.

It is more preferable that the raw materials contain 0.7% or less  $\text{Fe}_2\text{O}_3$  by weight.

It is more preferable that the combustible substance includes carbon.

It is more preferable that the talc, the aluminum hydroxide, and the fused silica occupy 80 weight% or more of the raw materials.

It is more preferable that the ratio in weight% of the large grains to the small grains in the aluminum hydroxide is in the range of 5/95 to 50/50.

It is more preferable to further comprise the step of drying the second mixture in the honeycomb shape.

Fig. 1 is a front view of a prior-art honeycomb structural body.

Fig. 2 is a sectional view taken along the line A-A in Fig. 1.

Fig. 3 is a diagram of a distribution of grains in aluminum hydroxide.

Fig. 4 is a diagram of the relation between mean pore diameter and porosity.

Fig. 5 is a diagram of the relation between pore diameter and cumulative pore volume.

Fig. 6 is a diagram of pressure losses provided by samples of honeycomb structural bodies.

Basic embodiments of this invention will be described hereinafter. A basic embodiment of this invention is directed to a honeycomb structural body having cordierite as a main component, the cordierite having a chemical composition including 45-55%  $\text{SiO}_2$ , 33-42%  $\text{Al}_2\text{O}_3$ , and 12-18%  $\text{MgO}$  by weight, the honeycomb structural body having a thermal expansion coefficient equal to or less than  $0.3 \times 10^{-6}/^\circ\text{C}$  from 25° C to 800° C, the honeycomb structural body having a porosity of 55-80%, the honeycomb structural body having a mean pore diameter of 25-40  $\mu\text{m}$ , the honeycomb structural body including partition walls, wherein pores at surfaces of the partition walls include small pores having diameters of 5-40  $\mu\text{m}$  and large pores having diameters of 40-100  $\mu\text{m}$ , and wherein the number of the small pores equals five to forty times that of the large pores.

The honeycomb structural body of the embodiment of this invention features that its thermal expansion coefficient is equal to or less than  $0.3 \times 10^{-6}/^\circ\text{C}$  from 25° C to 800° C, and its porosity is equal to 55-80%. The

honeycomb structural body of the embodiment of this invention also features that its mean pore diameter is equal to 25-40  $\mu\text{m}$ , and the number of the small pores equals five to forty times that of the large pores.

When the thermal expansion coefficient exceeds  $0.3 \times 10^{-6}/^\circ\text{C}$  from 25° C to 800° C, the heat resisting property of the honeycomb structural body tends to be poor.

When the porosity is smaller than 55%, a pressure loss provided by the honeycomb structural body tends to be high. It is preferable that the porosity is equal to or higher than 62%. When the porosity exceeds 80%, the filtration efficiency of the honeycomb structural body tends to be low. It is preferable that the porosity is equal to or less than 75%.

Regarding the pores at the surfaces of the partition walls, in the case where the number of the small pores is less than five times that of the large pores, the filtration efficiency tends to be low at an initial stage of a filtering process. In the case where the number of the small pores is greater than forty times that of the large pores, the filtration efficiency tends to be low and the product strength tends to be low.

The honeycomb structural body of the embodiment of this invention may be in the form of a cylinder having a plurality of axially-extending passages for introducing and discharging exhaust gas. The passages are defined by partition walls. Alternatively, the honeycomb structural body may be in the form of a rectangular parallelepiped, or may be in another shape. It is preferable that the passages have a quadrilateral cross-section. The passages may have a hexagonal cross-section, a triangular cross-section, or another cross-section.

Another basic embodiment of this invention is directed to a method of manufacturing a honeycomb structural body having cordierite as a main component, the cordierite having a chemical composition including 45-55%  $\text{SiO}_2$ , 33-42%  $\text{Al}_2\text{O}_3$ , and 12-18%  $\text{MgO}$  by weight, the honeycomb structural body having a thermal expansion coefficient equal to or less than  $0.3 \times 10^{-6}/^\circ\text{C}$  from 25° C to 800° C, the honeycomb structural body having a porosity of 55-80%, the honeycomb structural body having a mean pore diameter of 25-40  $\mu\text{m}$ , the honeycomb structural body including partition walls, wherein pores at surfaces of the partition walls include small pores having diameters of 5-40  $\mu\text{m}$  and large pores having diameters of 40-100  $\mu\text{m}$ , and wherein the number of the small pores equals five to forty times that of the large pores, the method comprising the steps of preparing talc containing 0.1-0.6%  $\text{Fe}_2\text{O}_3$  and 0.35% or less ( $\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O}$ ) by weight; preparing aluminum hydroxide containing small grains having diameters of 0.5-3  $\mu\text{m}$  and large grains having diameters of 5-15  $\mu\text{m}$ , the small grains and the large grains occupying 50-100% of all grains, wherein the ratio in weight% of the large grains to the small grains is in the range of 5/95 to 95/5; preparing fused silica containing 0.01% or less ( $\text{Na}_2\text{O} + \text{K}_2\text{O}$ ) by weight, and having a mean grain diameter of 30-100  $\mu\text{m}$ ; using the talc, the aluminum

hydroxide, and the fused silica as at least part of raw materials, and mixing the talc, the aluminum hydroxide, and the fused silica into a first mixture; adding at least one of organic blowing agent and combustible substance to the first mixture, the organic blowing agent starting to foam at 100° C or lower, the combustible substance starting to burn at lower than a sintering temperature, wherein the amount of added one of the organic blowing agent and the combustible substance is equal to 5-50 weight% with respect to the raw materials; kneading and mixing the first mixture and at least one of the organic blowing agent and combustible substance into a second mixture; making the second mixture into a honeycomb shape; drying the second mixture in the honeycomb shape; and sintering the second mixture in the honeycomb shape.

The method according to the embodiment of this invention features that the talc, the aluminum hydroxide, and the fused silica are used as raw materials. It is preferable that neither clay (for example, kaolin) nor alumina is positively used as raw material. The method according to the embodiment of this invention features that at least one of organic blowing agent and combustible substance is added to the raw materials.

When the  $\text{Fe}_2\text{O}_3$  content of the talc is less than 0.1 weight%, the range of sintering temperatures to provide low thermal expansion coefficients tends to be narrow. When the  $\text{Fe}_2\text{O}_3$  content of the talc exceeds 0.6 weight%, the thermal expansion coefficient tends to be great and the firing-based densification is promoted. Accordingly, in this case, it is difficult to provide a high porosity.

When the  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  content of the talc exceeds 0.35 weight%, the thermal expansion coefficient tends to be high.

As previously described, the aluminum hydroxide contains small grains having diameters of 0.5-3  $\mu\text{m}$  and large grains having diameters of 5-15  $\mu\text{m}$ . The method according to the embodiment of this invention features that the small grains and the large grains occupy 50-100% of all grains. In the case where the small grains and the large grains occupy less than 50% of all grains, it tends to be difficult to provide a desired porosity and a desired thermal expansion coefficient.

In the method according to the embodiment of this invention, the aluminum hydroxide may have a grain distribution as shown in Fig. 3. With reference to Fig. 3, the total number of small grains having diameters of 0.5-3  $\mu\text{m}$  and large grains having diameters of 5-15  $\mu\text{m}$  is greater than the total number of grains having other diameters.

Regarding the aluminum hydroxide, when the ratio in weight% of the large grains to the small grains is less than 5/95, the reactivity tends to be low and the thermal expansion coefficient tends to be great. It is preferable that the ratio in weight% of the large grains to the small grains is equal to 50/50 or less. When the ratio in weight% of the large grains to the small grains exceeds 95/5, it tends to be difficult to provide great pore vol-

umes. It is preferable that the ratio in weight% of the large grains to the small grains is in the range of 5/95 to 95/5. It is more preferable that the ratio in weight% of the large grains to the small grains is in the range of 5/95 to 50/50.

Regarding the fused silica, when its mean grain diameter is smaller than 30  $\mu\text{m}$ , the total number of large pores tends to be small at the surfaces of the partition walls. When its mean grain diameter exceeds 100  $\mu\text{m}$ , the strength of the honeycomb structural body tends to be low.

As previously described, at least one of organic blowing agent and combustible substance is added to the first mixture of the raw materials. The organic blowing agent is of the type starting to foam at 100° C or lower. The combustible substance is of the type starting to burn at lower than a sintering temperature. The amount of added one of the organic blowing agent and the combustible substance is equal to 5-50 weight% with respect to the raw materials. When the amount of added one of the organic blowing agent and the combustible substance is smaller than 5 weight%, the mean pore diameter tends to be excessively small. When the amount of added one of the organic blowing agent and the combustible substance is greater than 50 weight%, the mean pore diameter tends to be excessively great.

The organic blowing agent may use "Maikurosuhuea" produced by Japanese maker "Matsumoto Ushi-Seiyaku". To enable the blowing agent to foam during the drying step, it is preferable to select the organic blowing agent from among the types starting to foam at 100° C or lower in correspondence with a heating temperature which occurs during the drying step.

In the method according to the embodiment of this invention, the combustible substance may use carbon.

In the method according to the embodiment of this invention, the step of making the second mixture into a honeycomb shape may be implemented by adding water (or other substance) to the raw materials, kneading and mixing the water and the raw materials, and extruding the resultant mixture. In this case, the resultant mixture is extruded into a honeycomb shape and is then cut to form a honeycomb structural body having a desired size.

It is preferable to dry the honeycomb structural body to evaporate the water therefrom. The drying step may be implemented by heating the honeycomb structural body at a temperature of, for example, about 80-100° C. It is preferable to suitably set the length of a time for which the honeycomb structural body remains heated.

It is preferable to implement the sintering at a temperature of, for example, 1,300-1,500° C for 5-20 hours. It is preferable to set the sintering temperature in view of the size of the honeycomb structural body. It is preferable to set the time length of the continuation of the sintering in view of the size of the honeycomb structural body.

The drying step and the sintering step may be sep-

arate steps. Alternatively, the drying step and the sintering step may be combined into a process during which the heating temperature is continuously changed from the drying temperature to the sintering temperature.

It is preferable that the  $\text{Fe}_2\text{O}_3$  content of the raw materials is equal to or less than 0.7 weight%. In the case where the  $\text{Fe}_2\text{O}_3$  content of the raw materials exceeds 0.7 weight% while the  $\text{Fe}_2\text{O}_3$  content of the talc is in the range of 0.1-0.6 weight%, the thermal expansion coefficient tends to be great.

It is preferable that the talc, the aluminum hydroxide, and the fused silica occupy 80 weight% or more of the raw materials. When clay (for example, kaolin) and alumina occupy more than 20 weight% of the raw materials, it tends to be difficult to provide a high porosity.

As previously described, the honeycomb structural body of the embodiment of this invention has a thermal expansion coefficient equal to or less than  $0.3 \times 10^{-6}/^\circ\text{C}$  from  $25^\circ\text{C}$  to  $800^\circ\text{C}$ . Accordingly, the honeycomb structural body is excellent in heat resisting property. The honeycomb structural body does not break even when being repetitively exposed to a rapid temperature variation.

The honeycomb structural body of the embodiment of this invention has a porosity of 55-80% and a mean pore diameter of 25-40  $\mu\text{m}$ . In the honeycomb structural body of the embodiment of this invention, the number of the small pores equals five to forty times that of the large pores. Accordingly, the honeycomb structural body has a high filtration efficiency and provides only a low pressure loss.

Generally, a high porosity and a large mean pore diameter are advantageous in attaining a low pressure loss. The honeycomb structural body of the embodiment of this invention has both a high porosity and a large mean pore diameter. In general, as the porosity and the mean pore diameter are excessively increased, the filtration efficiency tends to be dropped. As previously described, in the honeycomb structural body of the embodiment of this invention, pores at surfaces of the partition walls include small pores having diameters of 5-40  $\mu\text{m}$  and large pores having diameters of 40-100  $\mu\text{m}$ . The ratio in number between the small pores and the large pores is limited to within a given range. Accordingly, the honeycomb structural body of the embodiment of this invention has a high filtration efficiency although its porosity is high and its mean pore diameter is large.

As previously described, the honeycomb structural body of the embodiment of this invention has a high filtration efficiency, a low pressure loss, and a low thermal expansion coefficient.

In the method according to the embodiment of this invention, the talc, the aluminum hydroxide, and the fused silica are used as raw materials. Neither clay (for example, kaolin) nor alumina is positively added to the raw materials. It should be noted that clay (for example, kaolin) or alumina causes a framework of a honeycomb structural body which has only a small number of pores.

In the method according to the embodiment of this invention, the fused silica and the aluminum hydroxide cause a framework of a honeycomb structural body which is more porous.

The use of the aluminum hydroxide causes the fact that water of hydration evaporates therefrom and hence many pores occur. The fused silica decomposes during the sintering process, and thus exhibits a volume reduction. The volume reduction causes many pores. Therefore, the method according to the embodiment of this invention causes a higher porosity.

The small grains in the aluminum hydroxide mainly cause pores having diameters of 0.3-0.7  $\mu\text{m}$  while the large grains therein mainly cause pores having diameters of 3-7  $\mu\text{m}$ . The generated pores can be controlled by suitably setting the ratio in weight% of the large grains to the small grains in the range of 5/95 to 95/5.

In the method according to the embodiment of this invention, at least one of organic blowing agent and combustible substance is added to the raw materials. It is preferable that the amount of added one of the organic blowing agent and the combustible substance is equal to 5-50 weight% with respect to the raw materials. The mean pore diameter in the honeycomb structural body can be controlled by adjusting the amount of added one of the organic blowing agent and the combustible substance. Accordingly, it is possible to optimize the mean pore diameter.

In the method according to the embodiment of this invention, the  $\text{Fe}_2\text{O}_3$  content of the talc is in the range of 0.1-0.6 weight%. In addition, the  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  content of the talc is 0.35 weight% or less. The  $(\text{Na}_2\text{O} + \text{K}_2\text{O})$  content of the fused silica is 0.01 weight% or less. Therefore, the thermal expansion coefficient of the honeycomb structural body can be equal to a very small value. Furthermore, it is possible to attain a higher porosity.

A specific embodiment of this invention will be described hereinafter. A honeycomb structural body according to an embodiment of this invention may be similar in mechanical structure to the prior-art honeycomb structural body in Figs. 1 and 2. The honeycomb structural body contains cordierite as a main component. The cordierite has a chemical composition such that  $\text{SiO}_2$  corresponds to 45-55 weight%;  $\text{Al}_2\text{O}_3$  corresponds to 33-42 weight%; and  $\text{MgO}$  corresponds to 12-18 weight%. The honeycomb structural body has a thermal expansion coefficient of no greater than  $0.3 \times 10^{-6}/^\circ\text{C}$  from  $25^\circ\text{C}$  to  $800^\circ\text{C}$ , and a porosity of 55-80%. A mean diameter of pores in the honeycomb structural body is in the range of 25-40  $\mu\text{m}$ . Pores at surfaces of partition walls in the honeycomb structural body are classified into small pores having diameters of 5-40  $\mu\text{m}$  and large pores having diameters of 40-100  $\mu\text{m}$ . The number of the small pores equals five to forty times the number of the large pores.

Ten samples of the honeycomb structural body of the embodiment of this invention were fabricated as inventive samples E1, E2, E3, E4, E5, E6, E7, E8, E9,

and E10. Five samples of a reference honeycomb structural body were fabricated as comparative samples C1, C2, C3, C4, and C5. Measurement was made as to a thermal expansion coefficient, a porosity, and a mean pore diameter of each of the inventive samples E1-E10 and the comparative samples C1-C5. Investigations were made as to a heat resisting property, a filtration efficiency, and a pressure loss of each of the inventive samples E1-E10 and the comparative samples C1-C5.

The inventive sample E1 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "A", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 50% small-grain aluminum hydroxide "A" and 50% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were mixed with each other. A suitable amount of water was added to the mixture of the raw materials and the additive materials. The water-added mixture of the raw materials and the additive materials was kneaded so that the raw materials, the additive materials, and water were mixed with each other. The resultant mixture of the raw materials, the additive materials, and water was extruded into a honeycomb structural member by a honeycomb extrusion molding machine. The honeycomb structural member was cut into a desired length. The resultant honeycomb structural member was heated and dried by an electronic oven so that 80% or more of water was evaporated therefrom. Then, the honeycomb structural member was continuously exposed to a flow of hot air for 12 hours. The hot air had a temperature of 80° C. Thus, the honeycomb structural member was further dried. Subsequently, the honeycomb structural member was sintered or fired at a temperature of 1,400° C for 20 hours. As a result, the honeycomb structural member was made into a honeycomb structural body corresponding to the inventive sample E1.

The inventive sample E2 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "A", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 95% small-grain aluminum hydroxide "A" and 5% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the inventive sample E2 by steps

similar to the steps in making the inventive sample E1.

The inventive sample E3 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "A", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 30% small-grain aluminum hydroxide "A" and 70% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the inventive sample E3 by steps similar to the steps in making the inventive sample E1.

The inventive sample E4 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "A", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 5% small-grain aluminum hydroxide "A" and 95% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the inventive sample E4 by steps similar to the steps in making the inventive sample E1.

The inventive sample E5 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "A", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 5% small-grain aluminum hydroxide "A" and 95% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 5 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the inventive sample E5 by steps similar to the steps in making the inventive sample E1.

The inventive sample E6 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "B", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 5% small-grain aluminum hydroxide "A" and 95% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon.



The blowing agent and the carbon in the additive materials were 10 weight% and 40 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the inventive sample E6 by steps similar to the steps in making the inventive sample E1.

The inventive sample E7 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "C", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 50% small-grain aluminum hydroxide "A" and 50% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the inventive sample E7 by steps similar to the steps in making the inventive sample E1.

The inventive sample E8 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "C", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 50% small-grain aluminum hydroxide "A" and 50% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 5 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the inventive sample E8 by steps similar to the steps in making the inventive sample E1.

The inventive sample E9 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "C", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 30% small-grain aluminum hydroxide "A" and 70% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the inventive sample E9 by steps similar to the steps in making the inventive sample E1.

The inventive sample E10 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "C", and 44% aluminum hydroxide by weight. The

aluminum hydroxide consisted of 5% small-grain aluminum hydroxide "A" and 95% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the inventive sample E10 by steps similar to the steps in making the inventive sample E1.

The comparative sample C1 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 39.9% talc "B", 18% fused silica "A", and 42.1% aluminum hydroxide by weight. The aluminum hydroxide consisted of 50% small-grain aluminum hydroxide "A" and 50% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the comparative sample C1 by steps similar to the steps in making the inventive sample E1.

The comparative sample C2 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 41.6% talc "A", 13.7% fused silica "A", 24.5% large-grain aluminum hydroxide "B", 10.1% alumina, and 10.1% kaolin by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the comparative sample C2 by steps similar to the steps in making the inventive sample E1.

The comparative sample C3 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "D", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 50% small-grain aluminum hydroxide "A" and 50% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 2 weight% and 20 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural

body corresponding to the comparative sample C3 by steps similar to the steps in making the inventive sample E1.

The comparative sample C4 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "C", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 50% small-grain aluminum hydroxide "A" and 50% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 0.2 weight% and 3 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the comparative sample C4 by steps similar to the steps in making the inventive sample E1.

The comparative sample C5 was fabricated as follows. Raw materials to form cordierite were prepared. The raw materials consisted of 38% talc "A", 18% fused silica "C", and 44% aluminum hydroxide by weight. The aluminum hydroxide consisted of 50% small-grain aluminum hydroxide "A" and 50% large-grain aluminum hydroxide "B" by weight. Additive materials were prepared. The additive materials contained combustible or inflammable substance. Specifically, the additive materials were blowing agent (foaming agent) and carbon. The blowing agent and the carbon in the additive materials were 15 weight% and 40 weight% relative to the raw materials, respectively. The raw materials and the additive materials were made into a honeycomb structural body corresponding to the comparative sample C5 by steps similar to the steps in making the inventive sample E1.

Table 1 lists the raw materials and the additive materials used in making the inventive samples E1-E10 and the comparative samples C1-C5.

The raw materials and the additive materials used in making the inventive samples E1-E10 and the comparative samples C1-C5 had compositions (components) and other features as follows. The talc "A" contained 62.1%  $\text{SiO}_2$ , 31.6%  $\text{MgO}$ , 0.19%  $\text{Al}_2\text{O}_3$ , 0.58%  $\text{Fe}_2\text{O}_3$ , and 0.18%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The talc "A" had a mean grain diameter of about 20  $\mu\text{m}$ . The talc "A" had an ignition loss (loss of ignition, or LOI) of 5.24% at a temperature of 1,000° C. The talc "B" contained 62.8%  $\text{SiO}_2$ , 30.3%  $\text{MgO}$ , 0.12%  $\text{Al}_2\text{O}_3$ , 3.00%  $\text{Fe}_2\text{O}_3$ , and 0.09%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The talc "B" had a mean grain diameter of about 20  $\mu\text{m}$ . The talc "B" had an ignition loss of 4.94% at a temperature of 1,000° C. The fused silica "A" contained 99.5%  $\text{SiO}_2$ , 0.02%  $\text{Al}_2\text{O}_3$ , 0.04%  $\text{Fe}_2\text{O}_3$ , and 0.006%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The fused silica "A" had a mean grain diameter of about 40  $\mu\text{m}$ . The fused silica "A" had an ignition loss of 0.30% at a temperature of 1,000° C. The fused silica "B" contained 99.5%  $\text{SiO}_2$ ,

0.02%  $\text{Al}_2\text{O}_3$ , 0.04%  $\text{Fe}_2\text{O}_3$ , and 0.006%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The fused silica "B" had a mean grain diameter of about 30  $\mu\text{m}$ . The fused silica "B" had an ignition loss of 0.30% at a temperature of 1,000° C. The fused silica "C" contained 99.5%  $\text{SiO}_2$ , 0.02%  $\text{Al}_2\text{O}_3$ , 0.04%  $\text{Fe}_2\text{O}_3$ , and 0.006%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The fused silica "C" had a mean grain diameter of about 100  $\mu\text{m}$ . The fused silica "C" had an ignition loss of 0.20% at a temperature of 1,000° C. The fused silica "D" contained 99.5%  $\text{SiO}_2$ , 0.02%  $\text{Al}_2\text{O}_3$ , 0.04%  $\text{Fe}_2\text{O}_3$ , and 0.006%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The fused silica "D" had a mean grain diameter of about 150  $\mu\text{m}$ . The fused silica "D" had an ignition loss of 0.20% at a temperature of 1,000° C. The small-grain aluminum hydroxide "A" contained 0.01%  $\text{SiO}_2$ , 99.6%  $\text{Al}(\text{OH})_3$ , 0.01%  $\text{Fe}_2\text{O}_3$ , and 0.36%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The small-grain aluminum hydroxide "A" had a mean grain diameter of about 1  $\mu\text{m}$ . The small-grain aluminum hydroxide "A" had an ignition loss of 34.1% at a temperature of 1,000° C. The large-grain aluminum hydroxide "B" contained 0.01%  $\text{SiO}_2$ , 99.8%  $\text{Al}(\text{OH})_3$ , 0.01%  $\text{Fe}_2\text{O}_3$ , and 0.16%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The large-grain aluminum hydroxide "B" had a mean grain diameter of about 10  $\mu\text{m}$ . The large-grain aluminum hydroxide "B" had an ignition loss of 34.2% at a temperature of 1,000° C. The alumina contained 0.03%  $\text{SiO}_2$ , 99.4%  $\text{Al}_2\text{O}_3$ , and 0.09%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The alumina had a mean grain diameter of about 1  $\mu\text{m}$ . The alumina had an ignition loss of 0.30% at a temperature of 1,000° C. The kaolin contained 45.1%  $\text{SiO}_2$ , 38.5%  $\text{Al}_2\text{O}_3$ , 0.34%  $\text{Fe}_2\text{O}_3$ , and 0.20%  $(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$  by weight. The kaolin had a mean grain diameter of about 10  $\mu\text{m}$ . The kaolin had an ignition loss of 14.1% at a temperature of 1,000° C. The blowing agent had a mean grain diameter of about 15  $\mu\text{m}$ . The blowing agent had an ignition loss of 99.9% at a temperature of 1,000° C. The carbon had a mean grain diameter of about 50  $\mu\text{m}$ . The carbon had an ignition loss of 99.9% at a temperature of 1,000° C.

Regarding the raw materials and the additive materials used in making the inventive samples E1-E10 and the comparative samples C1-C5, Table 2 lists the components, the ignition losses (the losses of ignition or LOI), and the mean grain diameters.

Measurements were made as to a thermal expansion coefficient, a porosity, and a mean pore diameter of each of the inventive samples E1-E10 and the comparative samples C1-C5. The measurements of the thermal expansion coefficients were executed by a thermodilatometer. The measurements of the porosities and the mean pore diameters were executed in a method of mercury penetration which used a porosimeter to detect pore volumes.

Fig. 4 shows the measured mean pore diameters and the measured porosities of the inventive samples E1-E10 and the comparative samples C1-C5. In Fig. 4, the abscissa denotes mean pore diameter while the ordinate denotes porosity. In Fig. 4, the black small circles indicate the measured mean pore diameters and



the measured porosities of the inventive samples E1-E10 while the white small circles indicate the measured mean pore diameters and the measured porosities of the comparative samples C1-C5. With reference to Fig. 4, all the inventive samples E1-E10 are in a preferable range 92 of mean pore diameter and porosity. Further, some of the inventive samples E1-E10 are in a most preferable range 84 of mean pore diameter and porosity. On the other hand, all the comparative samples C1-C5 are outside the preferable range 92 of mean pore diameter and porosity.

Evaluations were made as to a heat resisting property, a filtration efficiency, and a pressure loss of each of the inventive samples E1-E10 and the comparative samples C1-C5. The heat resisting property corresponds to sample's ability to endure stress caused by thermal expansion. The heat resisting properties of the inventive samples E1-E10 and the comparative samples C1-C5 were evaluated on the basis of the measured thermal expansion coefficients thereof. Regarding the heat resisting properties, samples having thermal expansion coefficients of greater than  $0.3 \times 10^{-6}/^{\circ}\text{C}$  were concluded to be unacceptable while samples having thermal expansion coefficients equal to or smaller than  $0.3 \times 10^{-6}/^{\circ}\text{C}$  were concluded to be acceptable. The filtration efficiencies of the inventive samples E1-E10 and the comparative samples C1-C5 were evaluated as follows. A filter using each of the samples was prepared. Gas containing particulates was past through the filter. Measurement was made as to the rate of a reduction in amount of particulates which was caused by the filter. Regarding the filtration efficiencies, samples corresponding to particulate reduction rates of greater than 80% were concluded to be acceptable while samples corresponding to particulate reduction rates equal to or smaller than 80% were concluded to be unacceptable. The pressure losses of the inventive samples E1-E10 and the comparative samples C1-C5 were evaluated as follows. A filter using each of the samples was prepared. A diesel engine having a displacement of 2,200 cc was prepared. The filter remained in an exhaust passage of the diesel engine for 4 hours during which the engine continued to be operated at an engine speed of 2,000 rpm and a torque of 100 Nm. Then, a pressure loss provided by the filter was measured. Samples corresponding to pressure losses of 10 KPa or less were concluded to be acceptable while samples corresponding to pressure losses of greater than 10 KPa were concluded to be unacceptable.

The measured thermal expansion coefficient of the inventive sample E1 was equal to  $0.15 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E1 was equal to 58.0%. The measured mean pore diameter of the inventive sample E1 was equal to about 27.0  $\mu\text{m}$ . The inventive sample E1 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E1 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the inventive sample E2 was equal to  $0.15 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E2 was equal to 55.2%. The measured mean pore diameter of the inventive sample E2 was equal to about 25.5  $\mu\text{m}$ . The inventive sample E2 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E2 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the inventive sample E3 was equal to  $0.16 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E3 was equal to 62.0%. The measured mean pore diameter of the inventive sample E3 was equal to about 28.2  $\mu\text{m}$ . The inventive sample E3 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E3 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the inventive sample E4 was equal to  $0.18 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E4 was equal to 63.6%. The measured mean pore diameter of the inventive sample E4 was equal to about 29.0  $\mu\text{m}$ . The inventive sample E4 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E4 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the inventive sample E5 was equal to  $0.22 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E5 was equal to 71.0%. The measured mean pore diameter of the inventive sample E5 was equal to about 39.6  $\mu\text{m}$ . The inventive sample E5 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E5 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the inventive sample E6 was equal to  $0.23 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E6 was equal to 80.0%. The measured mean pore diameter of the inventive sample E6 was equal to about 26.0  $\mu\text{m}$ . The inventive sample E6 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E6 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the inventive sample E7 was equal to  $0.28 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E7 was equal to 59.5%. The measured mean pore diameter of the inventive sample E7 was equal to about 36.1  $\mu\text{m}$ . The inventive sample E7 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E7 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the inventive sample E8 was equal to  $0.28 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E8 was equal to 63.1%. The measured mean pore diameter of the inventive sample E8 was equal to about 38.1  $\mu\text{m}$ .

The inventive sample E8 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E8 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the inventive sample E9 was equal to  $0.29 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E9 was equal to 72.6%. The measured mean pore diameter of the inventive sample E9 was equal to about 30.1  $\mu\text{m}$ . The inventive sample E9 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E9 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the inventive sample E10 was equal to  $0.29 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the inventive sample E10 was equal to 75.0%. The measured mean pore diameter of the inventive sample E10 was equal to about 37.5  $\mu\text{m}$ . The inventive sample E10 was concluded to be acceptable in all of heat resisting property, filtration efficiency, and pressure loss. Thus, the inventive sample E10 was found to be eligible by synthetic judgment.

The measured thermal expansion coefficient of the comparative sample C1 was equal to  $0.19 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the comparative sample C1 was equal to 53.9%. The measured mean pore diameter of the comparative sample C1 was equal to about 22.2  $\mu\text{m}$ . The comparative sample C1 was concluded to be acceptable in heat resisting property and filtration efficiency. The comparative sample C1 was concluded to be unacceptable in pressure loss. Thus, the comparative sample C1 was found to be ineligible by synthetic judgment.

The measured thermal expansion coefficient of the comparative sample C2 was equal to  $0.20 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the comparative sample C2 was equal to 48.3%. The measured mean pore diameter of the comparative sample C2 was equal to about 21.3  $\mu\text{m}$ . The comparative sample C2 was concluded to be acceptable in heat resisting property and filtration efficiency. The comparative sample C2 was concluded to be unacceptable in pressure loss. Thus, the comparative sample C2 was found to be ineligible by synthetic judgment.

The measured thermal expansion coefficient of the comparative sample C3 was equal to  $0.32 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the comparative sample C3 was equal to 60.6%. The measured mean pore diameter of the comparative sample C3 was equal to about 40.2  $\mu\text{m}$ . The comparative sample C3 was concluded to be acceptable in filtration efficiency and pressure loss. The comparative sample C3 was concluded to be unacceptable in heat resisting property. Thus, the comparative sample C3 was found to be ineligible by synthetic judgment.

The measured thermal expansion coefficient of the comparative sample C4 was equal to  $0.28 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the comparative sample C4 was equal to 52.0%. The measured mean pore diame-

ter of the comparative sample C4 was equal to about 19.4  $\mu\text{m}$ . The comparative sample C4 was concluded to be acceptable in heat resisting property and filtration efficiency. The comparative sample C4 was concluded to be unacceptable in pressure loss. Thus, the comparative sample C4 was found to be ineligible by synthetic judgment.

The measured thermal expansion coefficient of the comparative sample C5 was equal to  $0.29 \times 10^{-6}/^{\circ}\text{C}$ . The measured porosity of the comparative sample C5 was equal to 67.1%. The measured mean pore diameter of the comparative sample C5 was equal to about 52.0  $\mu\text{m}$ . The comparative sample C5 was concluded to be acceptable in heat resisting property and pressure loss. The comparative sample C5 was concluded to be unacceptable in filtration efficiency. Thus, the comparative sample C5 was found to be ineligible by synthetic judgment.

Regarding the inventive samples E1-E10 and the comparative samples C1-C5, Table 3 lists the thermal expansion coefficients, the porosities, the mean pore diameters, the results of judgment as to the heat resisting properties, the results of the judgment as to the filtration efficiency, the results of judgment as to the pressure losses, and the results of the synthetic judgment.

It is thought that since the comparative sample C1 uses talc having a high  $\text{Fe}_2\text{O}_3$  content, the porosity and the mean pore diameter thereof are small. It is thought that the small porosity and the small mean pore diameter cause a high pressure loss provided by the comparative sample C1.

It is thought that since the comparative sample C2 uses more than 20 weight% kaolin and alumina as raw materials, the porosity and the mean pore diameter thereof are small. It is thought that the small porosity and the small mean pore diameter cause a high pressure loss provided by the comparative sample C2.

It is thought that since the comparative sample C3 uses fused silica having a large mean grain diameter (about 150  $\mu\text{m}$ ), the mean pore diameter and the thermal expansion coefficient thereof are great. It is thought that the great mean pore diameter and the great thermal expansion coefficient cause a poor heat resisting property of the comparative sample C3.

It is thought that since blowing agent and carbon (additive materials) used for the comparative sample C4 are small in weight% (3.2 weight%) relative to the raw materials, the mean pore diameter thereof is small. It is thought that the small mean pore diameter causes a high pressure loss provided by the comparative sample C4.

It is thought that since blowing agent and carbon (additive materials) used for the comparative sample C5 are great in weight% (55 weight%) relative to the raw materials, the mean pore diameter thereof is large. It is thought that the large mean pore diameter causes a low filtration efficiency of the comparative sample C5.

Fig. 5 shows the results of porosimeter-based

measurement of cumulative pore volumes relative to pore diameter in the inventive sample E1 and the comparative sample C2. It is shown in Fig. 5 that the cumulative pore volume in the inventive sample E1 is remarkably greater than that in the comparative sample C2. Accordingly, it is clear that the porosity of the inventive sample E1 is higher than that of the comparative sample C2.

Fig. 6 shows pressure losses provided by the inventive sample E1 and the comparative sample C2 respectively. It is shown in Fig. 6 that the pressure loss provided by the inventive sample E1 is smaller than that by the comparative sample C2 by about 30%.

#### Claims

1. A honeycomb structural body having cordierite as a main component, the cordierite having a chemical composition including 45-55% SiO<sub>2</sub>, 33-42% Al<sub>2</sub>O<sub>3</sub>, and 12-18% MgO by weight, the honeycomb structural body having a thermal expansion coefficient equal to or less than  $0.3 \times 10^{-6}/^{\circ}\text{C}$  from 25° C to 800° C, the honeycomb structural body having a porosity of 55-80%, the honeycomb structural body having a mean pore diameter of 25-40 μm, the honeycomb structural body including partition walls, wherein pores at surfaces of the partition walls include small pores having diameters of 5-40 μm and large pores having diameters of 40-100 μm, and wherein the number of the small pores equals five to forty times that of the large pores.
2. A honeycomb structural body as recited in claim 1, wherein the porosity is in the range of 62-75%.
3. A method of manufacturing a honeycomb structural body having cordierite as a main component, the cordierite having a chemical composition including 45-55% SiO<sub>2</sub>, 33-42% Al<sub>2</sub>O<sub>3</sub>, and 12-18% MgO by weight, the honeycomb structural body having a thermal expansion coefficient equal to or less than  $0.3 \times 10^{-6}/^{\circ}\text{C}$  from 25° C to 800° C, the honeycomb structural body having a porosity of 55-80%, the honeycomb structural body having a mean pore diameter of 25-40 μm, the honeycomb structural body including partition walls, wherein pores at surfaces of the partition walls include small pores having diameters of 5-40 μm and large pores having diameters of 40-100 μm, and wherein the number of the small pores equals five to forty times that of the large pores, the method comprising the steps of:

preparing talc containing 0.1-0.6% Fe<sub>2</sub>O<sub>3</sub> and 0.35% or less (CaO + Na<sub>2</sub>O + K<sub>2</sub>O) by weight; preparing aluminum hydroxide containing small grains having diameters of 0.5-3 μm and large grains having diameters of 5-15 μm, the small grains and the large grains occupying 50-100%

of all grains, wherein the ratio in weight% of the large grains to the small grains is in the range of 5/95 to 95/5;

preparing fused silica containing 0.01% or less (Na<sub>2</sub>O + K<sub>2</sub>O) by weight, and having a mean grain diameter of 30-100 μm;

using the talc, the aluminum hydroxide, and the fused silica as at least part of raw materials, and mixing the talc, the aluminum hydroxide, and the fused silica into a first mixture;

adding at least one of organic blowing agent and combustible substance to the first mixture, the organic blowing agent starting to foam at 100° C or lower, the combustible substance starting to burn at lower than a sintering temperature, wherein the amount of said added one of the organic blowing agent and the combustible substance is equal to 5-50 weight% with respect to the raw materials;

kneading and mixing the first mixture and said at least one of the organic blowing agent and combustible substance into a second mixture; making the second mixture into a honeycomb shape; and

sintering the second mixture in the honeycomb shape.

4. A method as recited in claim 3, wherein the raw materials contain 0.7% or less Fe<sub>2</sub>O<sub>3</sub> by weight.
5. A method as recited in claim 3 or claim 4, wherein the combustible substance includes carbon.
6. A method as recited in claim 3, wherein the talc, the aluminum hydroxide, and the fused silica occupy 80 weight% or more of the raw materials.
7. A method as recited in claim 3, wherein the ratio in weight% of the large grains to the small grains in the aluminum hydroxide is in the range of 5/95 to 50/50.
8. A method as recited in claim 3, further comprising the step of drying the second mixture in the honeycomb shape.

TABLE 1

SAMPLE NO.	CORDIERITE RAW MATERIALS (WEIGHT %)					WEIGHT % TO RAW MATERIALS	
	TALC	FUSED SILICA	ALUMINUM HYDROXIDE SMALL : LARGE GRAIN : GRAIN	ALUMINA	KAOLIN	BLOWING AGENT	CARBON
E1	TALC A 38%	SILICA A 18%	50 : 50 44%	—	—	2%	20%
E2	TALC A 38%	SILICA A 18%	95 : 5 44%	—	—	↑	↑
E3	↑	↑	30 : 70 44%	—	—	↑	↑
E4	↑	↑	5 : 95 44%	—	—	↑	↑
E5	↑	↑	↑	—	—	5%	20%
E6	↑	SILICA B 18%	↑	—	—	10%	40%
E7	↑	SILICA C 18%	50 : 50 44%	—	—	2%	20%
E8	↑	↑	↑	—	—	5%	↑
E9	↑	↑	30 : 70 44%	—	—	2%	↑
E10	↑	↑	5 : 95 44%	—	—	↑	↑
C1	TALC B 39.9%	SILICA A 18%	50 : 50 42.1%	—	—	2%	20%
C2	TALC A 41.6%	SILICA A 13.7%	ALUMINUM HYDROXIDE B 24.5%	ALUMINA 10.1%	KAOLIN 10.1%	2%	20%
C3	TALC A 38%	SILICA D 18%	50 : 50 44%	—	—	2%	20%
C4	↑	SILICA C 18%	50 : 50 44%	—	—	0.2%	3%
C5	↑	↑	50 : 50 44%	—	—	15%	40%

TABLE 2

MATERIAL NAME	MEAN GRAIN DIAMETER	SiO <sub>2</sub>	NgO	Al <sub>2</sub> O <sub>3</sub>	Al (OH) <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	CaO + Na <sub>2</sub> O + K <sub>2</sub> O	LOI
TALC A	20 $\mu$ m	62.1	31.6	0.29	—	0.58	0.18	5.24
TALC A	20 $\mu$ m	62.8	30.3	0.12	—	3.00	0.09	4.94
FUSED SILICA A	40 $\mu$ m	99.5	—	0.02	—	0.04	0.006	0.30
FUSED SILICA B	30 $\mu$ m	99.5	—	0.02	—	0.04	0.006	0.30
FUSED SILICA C	100 $\mu$ m	99.5	—	0.02	—	0.04	0.006	0.20
FUSED SILICA D	150 $\mu$ m	99.5	—	0.02	—	0.04	0.006	0.20
ALUMINUM HYDROXIDE A	1 $\mu$ m	0.01	—	—	99.6	0.01	0.36	34.1
ALUMINUM HYDROXIDE B	10 $\mu$ m	0.01	—	—	99.8	0.01	0.16	34.2
ALUMINA	1 $\mu$ m	0.03	—	99.4	—	—	0.09	0.30
KAOLIN	10 $\mu$ m	45.1	—	38.5	—	0.34	0.20	14.1
BLOWING AGENT	15 $\mu$ m	—	—	—	—	—	—	99.9
CARBON	50 $\mu$ m	—	—	—	—	—	—	99.9

TABLE 3

SAMPLE NO.	FILTER PROPERTIES			JUDGMENT			SYNTHETIC JUDGMENT
	THERMAL EXPANSION COEFFICIENT $\times 10^{-6}/^{\circ}\text{C}$	POROSITY %	MEAN PORE DIAMETER $\mu\text{m}$	HEAT RESISTING PROPERTY	FILTRATION EFFICIENCY	PRESSURE LOSS	
E1	0.15	58.0	27.0	○	○	○	○
E2	0.15	55.2	25.5	○	○	○	○
E3	0.16	62.0	28.2	○	○	○	○
E4	0.18	63.6	29.0	○	○	○	○
E5	0.22	71.0	39.6	○	○	○	○
E6	0.23	80.0	26.0	○	○	○	○
E7	0.28	59.5	36.1	○	○	○	○
E8	0.28	63.1	38.1	○	○	○	○
E9	0.29	72.6	30.1	○	○	○	○
E10	0.29	75.0	37.5	○	○	○	○
C1	0.19	53.9	22.2	○	○	×	×
C2	0.20	48.3	21.3	○	○	×	×
C3	0.32	60.6	40.2	×	○	○	×
C4	0.28	52.0	19.4	○	○	×	×
C5	0.29	67.1	52.0	○	×	○	×



FIG. 1

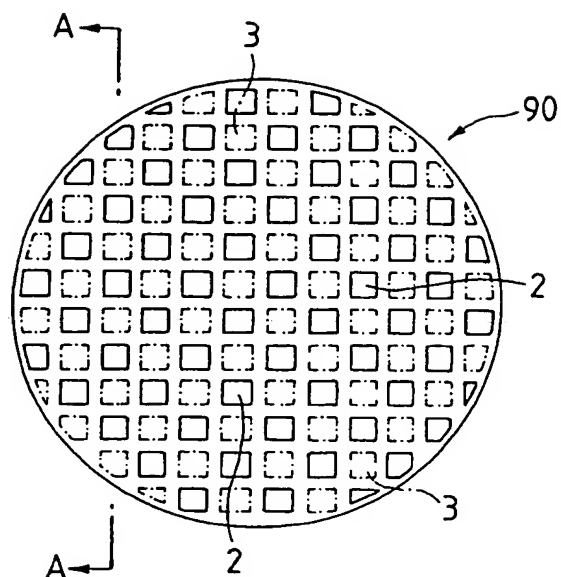


FIG. 2

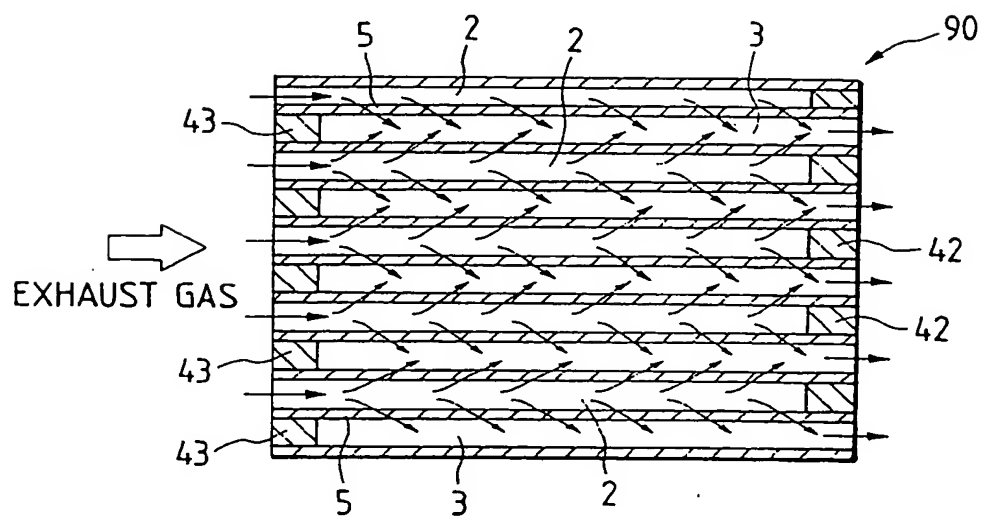


FIG. 3

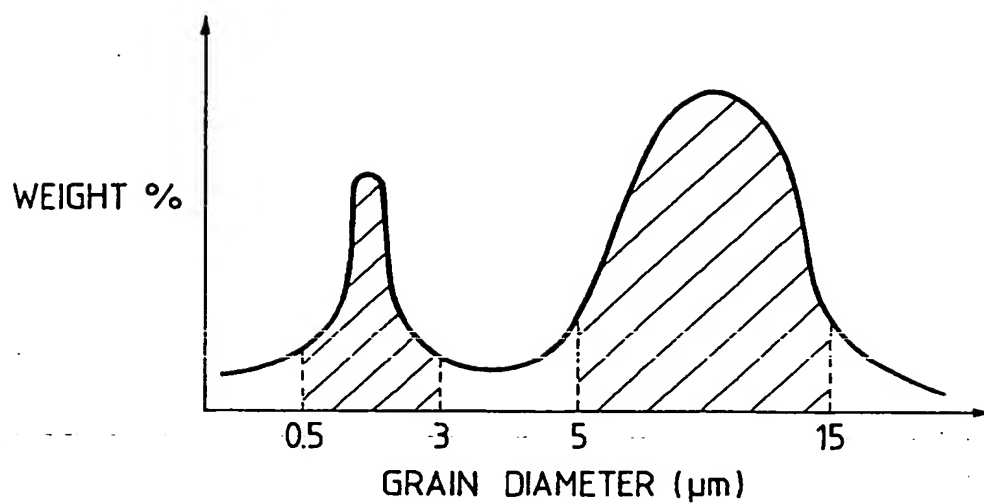


FIG. 4

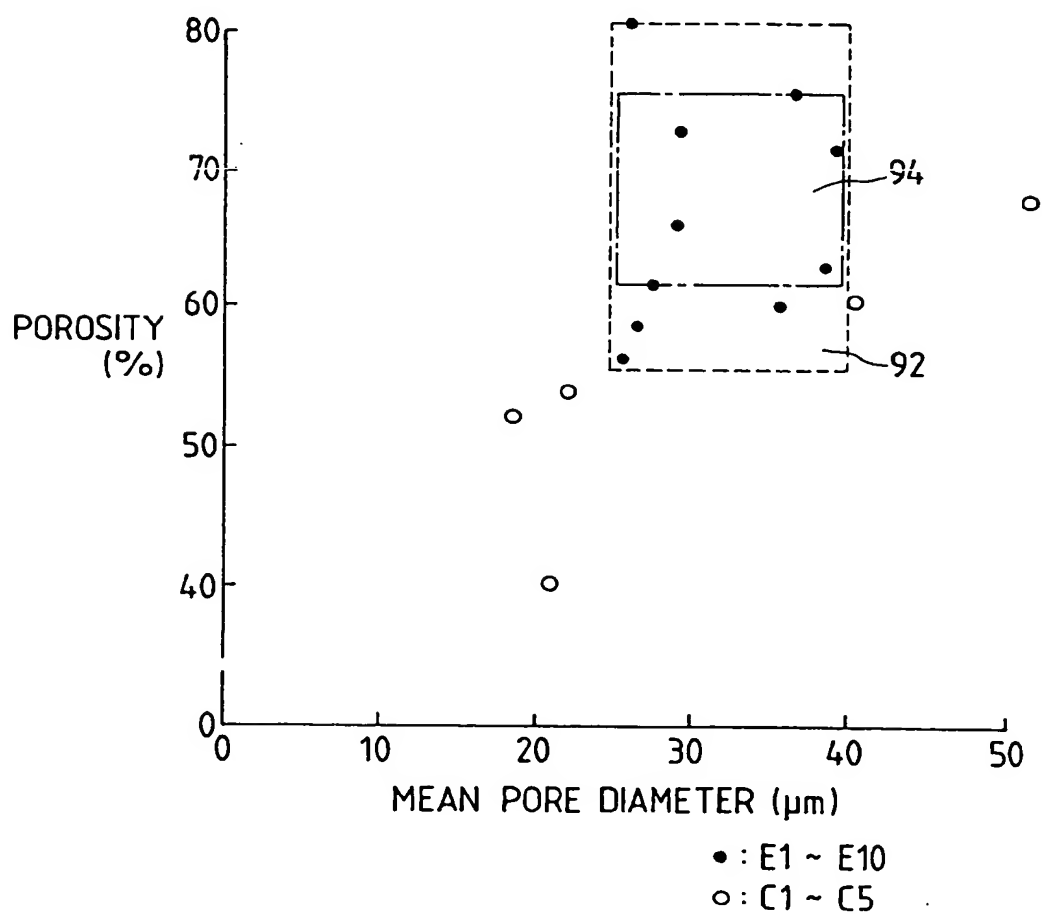


FIG. 5

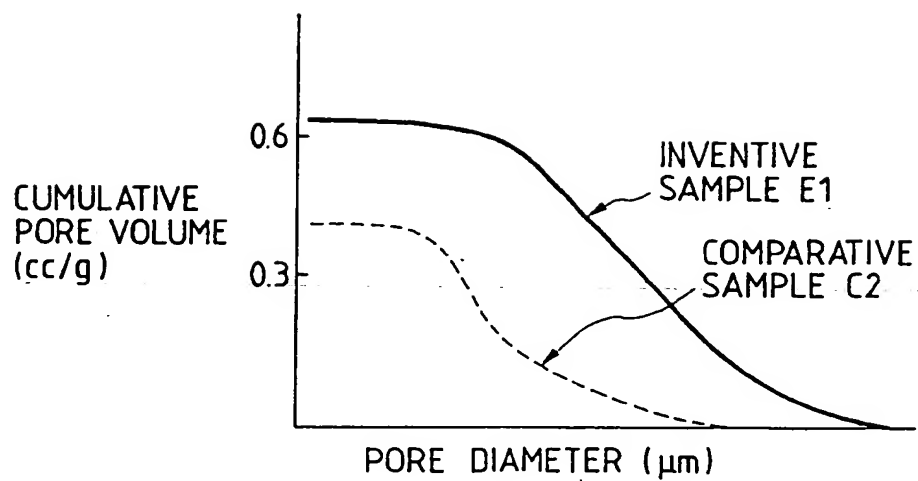
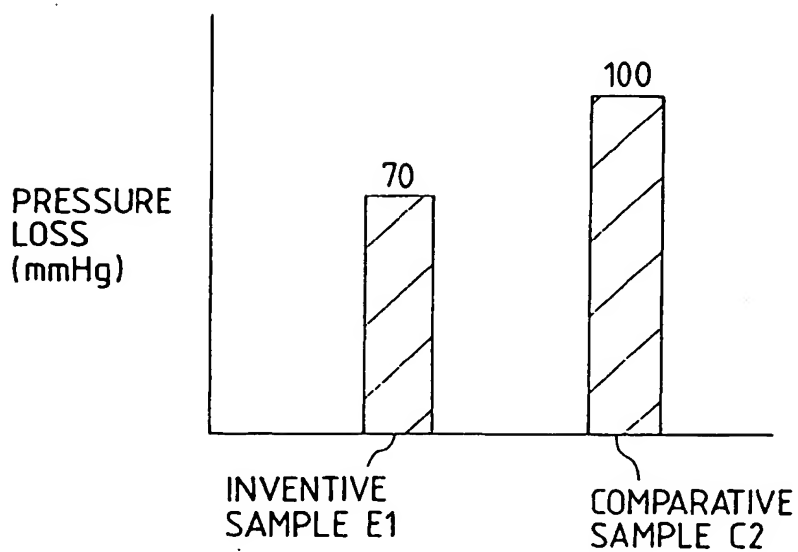


FIG. 6





European Patent  
Office

# EUROPEAN SEARCH REPORT

Application Number  
EP 96 11 1211

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
X	DE 35 41 372 A (NIPPON DENSO CO) 5 June 1986	1,2	C04B35/195 C04B38/00
A	* claims; figure 4; examples 1-5; table 1 *	3-8	
A	US 4 416 675 A (MONTIERTH MAX R) 22 November 1983 * claims *	1	
A	EP 0 232 621 A (NGK INSULATORS LTD) 19 August 1987 * claims; tables *	1-8	
A	EP 0 278 749 A (NGK INSULATORS LTD) 17 August 1988 * claims *	1-8	
A	EP 0 549 873 A (CORNING INC) 7 July 1993 * claims; examples *	1-8	
A	EP 0 545 008 A (CORNING INC) 9 June 1993 * claims *	1-8	TECHNICAL FIELDS SEARCHED (Int.Cl.6)
D	& US 5 258 150 A		C04B
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A	EP 0 658 363 A (NGK INSULATORS LTD) 21 June 1995 * the whole document *	1-8	
The present search report has been drawn up for all claims			
Place of search BERLIN		Date of completion of the search 21 October 1996	Examiner Kuehne, H-C
<b>CATEGORY OF CITED DOCUMENTS</b> X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	

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